# Studies on oxidized **low** density lipoproteins. Controlled oxidation and a prostaglandin artifact

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Abstract Low density lipoproteins (LDL), isolated by ultracentrifugal flotation, were oxidized  $(LDL<sub>OXID</sub>)$  slowly during dialysis against 0.15 **M** NaCl and subsequent incubation in 96% air-4% CO<sub>2</sub> at 37°C. Butylated hydroxytoluene prevented LDL oxidation. LDL preparations from different sera were oxidized at different rates and the degree of lipid peroxidation was controlled by varying the incubation time. Mild oxidation did not alter the electrophoretic mobility of the  $LDL<sub>OXID</sub>$  preparation. LDLOXID contained lipid peroxides in neutral lipids, had increased amounts of lysophosphatidylcholine, and contained a number of complex oxidation products that were generated from the oxidation of free fatty acids. These oxidation products included large amounts of soluble material that cross-reacted with antibodies to PGE, but not 6-keto-PGF $_{10}$ . The amount of cross-reacting material was proportional to the degree of lipid peroxidation. Cross-reacting material in LDLOXID preparations was evidently formed from the oxidation of free fatty acids released from LDL, since cross-reacting material was also formed when a synthetic fat emulsion was oxidized in the presence of free arachidonic acid.-Zhang, **H.,** W. **B. Davis, X.** Chen, **R. L.** Whisler, and **D.** *G.* Cornwell. Studies on oxidized low density lipoproteins. Controlled oxidation and a prostaglandin artifact. *J Lipid Res.* 1989. *30:* 141-148.

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The potential role of lipid peroxidation in the pathogenesis of atherosclerosis continues to be an area of considerable interest and controversy (1-4). The susceptibility of isolated plasma low density lipoproteins (LDL) to lipid peroxidation was recognized over *30* years ago (5, 6) and recent studies (7-10) have identified lipid peroxides as malondialdehyde (MDA) in serum **(3** to **7** nmol MDA/ml), and LDL (3 to 15 nmol MDA/mg LDL cholesterol). A number of studies describe specific effects of oxidized low density lipoproteins (LDL $_{\rm OXID}$ ) that may be related to atherosclerosis.  $LDL<sub>OXID</sub>$  are taken up by the scavenger receptor of the monocyte/macrophage and degraded by these cells more rapidly than native LDL (II-Zl),

We have recently begun a series of studies (22) with  $LDL<sub>OXID</sub>$  that were prepared by controlled mild oxidation. The  $LDL<sub>OXID</sub>$  preparations were characterized by differences in lipid peroxide content that were measured as thiobarbituric acid reactants (TBAR). LDL<sub>OXID</sub> apparently contained lipid oxidation products with crossreactivity in the radioimmunoassay (RIA) of prostanoids. Studies on the formation and properties of  $LDL<sub>OXID</sub>$  and an oxidized synthetic fat emulsion Intralipid $^{\circledast}$  (II) are presented in this report.

# MATERIALS AND METHODS

# Materials

Arachidonic acid (AA) was purchased from Nu-Chek-Prep (Elysian, MN), purified by elution from a Unisil<sup>®</sup> column with hexane-ether 9:l (v/v), and used only when thin-layer chromatography (TLC) showed that lipid peroxides were absent (23). Other reagents were obtained Studies on the formation and properties of LDL<sub>OXID</sub> and<br>an oxidized synthetic fat emulsion Intralipid<sup>®</sup> (II) are<br>presented in this report.<br>MATERIALS AND METHODS<br>Materials<br>Arachidonic acid (AA) was purchased from Nu-Chek-(200.0 Ci/mmol), 6-[5,8,9,11,12,13,15-<sup>3</sup>H(N)]-keto-PGF<sub>1a</sub> (163.5 Ci/mmol), and  $[U^{-14}C]AA$  (1.0 Ci/mmol) (New England Nuclear, Boston, MA); phosphatidylinositol (PI) and phosphatidylserine (PS) (Avanti Polar-Lipids, Inc., Birmingham, AL); phosphatidic acid (PA), lysophosphatidylcholine (LPC), phosphatidylcholine (PC), and phosphatidylethanolamine (PE) (Sigma Chemical *Go.,*  St. Louis, MO); sphingomyelin (Sph) (Serdary, London,

**Abbreviations: AA, arachidonic acid; BHT, butylated hydroxytoluene; C, cholesterol; CE, cholesteryl ester; FFA, free fatty acid; HPLC, high performance liquid chromatography; Il, Intralipid<sup>®</sup>; Il<sub>OXID</sub>, oxidized Il;** LDL, low density lipoproteins; LDL<sub>BHT</sub>, unoxidized LDL; LDL<sub>OXID</sub>, **oxidized LDL; LPC, lysophosphatidylcholine; MDA, malondialdehyde; NL, neutral lipid; PA, phosphatidic acid; PC, phosphatidylcholine; PE, phosphatidylethanolamine; PI, phosphatidylinositol; PS, phosphatidylserine; RIA, radioimmunoassay; Sph, sphingomyelin; TBAR, thiobarbituric acid reactants;** TG, **triglycerides; TLC, thin-layer chromatography; TNS, 6-p-toluidino-2-naphthalenesulfonic acid; IM, indomethacin.** 

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Ontario, Canada); trilinolenin (Nu-Chek-Prep); butylated hydroxytoluene (BHT), and 1,1,3,3 -tetramethoxypropane (Aldrich, Milwaukee, WI); I1 (KabiVitrum, Alameda, CA). Tissue culture media was previously described  $(22-24)$ .

# **LDL preparations**

Individual units of freshly drawn human plasma were obtained from the Red Cross and converted to serum by the addition of CaCl<sub>2</sub>. LDL (d 1.019-1.063 g/ml) were isolated by ultracentrifugation (6). A portion of the LDL preparation was dialyzed at  $4^{\circ}$ C for 16 hr against 400 vol of 0.15 M NaCl and then sterilized by passage through a  $0.45$ - $\mu$ m Millipore filter. Total cholesterol was measured by an established procedure (25); LDL concentrations are reported as  $\mu$ g cholesterol throughout this study.  $LDL<sub>OXID</sub>$  were prepared by incubating sterilized LDL (ca. 4,000  $\mu$ g cholesterol/ml) at 37°C in 96% air-4% CO<sub>2</sub> for various time intervals. The oxidation reaction was monitored by the disappearance of the yellow carotenoid color (6) and the appearance of TBAR. In some experiments, 10  $\mu$ M indomethacin (IM) was added before oxidation.  $LDL_{BHT}$  were prepared by the immediate addition of 100  $\mu$ M BHT to a second portion of the fresh LDL preparation before dialysis. This mixture was dialyzed for 16 hr against 400 vol of 0.15 M NaCl saturated with BHT and then sterilized by filtration.  $LDL<sub>BHT</sub>$  were incubated at 37°C for the same number of hours as  $LDL<sub>OXID</sub>$ .  $LDL<sub>OXID</sub>$  and  $LDL<sub>BHT</sub>$  were further characterized by ultracentrifugal flotation (6) and electrophoresis in agarose gel (26).

## **Oxidized Intralipid (Iloxin)**

Intralipid diluted in saline, unlike LDL, was not oxidized during incubation at 37°C in 96% air-4%  $CO<sub>2</sub>$ . However, large amounts of lipid peroxides (TBAR) were formed when 10  $\mu$ M  $\text{Fe}^{3+}$  was added to the incubation system. Il $_{\rm OXID}$  containing different amounts of lipid peroxides were obtained by varying the incubation times.

# **Separation of lipid classes**

Lipids were extracted from LDL $_{\text{BHT}}$  and LDL $_{\text{OXID}}$ with chloroform-methanol 2:1 (v/v) or  $LDL_{BHT}$  and  $LDL<sub>OXID</sub>$  were applied directly to a TLC plate. Phospholipid classes were separated by TLC on Whatman LK5D plates. The developing solvent consisted of chloroform-methanol-40% methylamine  $60:20:5$  (v/v) (27). Phospholipid standards PI, LPC, Sph, PC, and PE were well separated. PS was not separated from PA. Neutral lipids moved near the solvent front and free AA separated as a distinct spot  $(R_f 0.75)$  in this system.

Neutral lipid (NL) classes were separated by TLC on Whatman LK6D plates. The developing solvent consisted of hexane-ether-acetic acid 80:20:1 (v/v). Cholesteryl

**142 Journal of Lipid Research Volume 30, 1989** 

esters (CE), triglycerides (TG), cholesterol (C), and free AA were well separated. Phospholipids remained at the origin in this system.

Lipid classes were visualized with iodine vapor or visualized under UV light after spraying with 1 mM 6-p**toluidino-2-naphthalenesulfonic** acid (TNS) in buffer solution (28). Lipid peroxides were visualized by a starch-iodide spray (23).

High performance liquid chromatography (HPLC) was used as previously described (24) to separate labeled AA derivatives formed by oxidation. We used a Beckman model 334 chromatograph which consisted of dual Beckman Model 110A pumps (A and B), Altex 210A injector with a 20-µl loop, Beckman controller, Beckman 164 variable wavelength detector, and an Altex Model C-RIA integrator. HPLC was performed at room temperature on a 250  $\times$  4.6 mm Ultrasphere-ODS (5  $\mu$ m) reversed-phase column preceded by  $35 \times 4.6$  mm guard column packed with ODs.

The elution solvents for HPLC consisted of various mixtures of acetonitrile-isopropanol-aqueous phosphoric acid (pH 2). Solvent compositions were: solvent A, HPLC grade water adjusted to pH 2 by phosphoric acid; solvent  $B_1$ , acetonitrile; solvent  $B_2$ , 3% isopropanol in acetonitrile. The flow rate of elution solvents was 1.0 ml/min. The elution steps were **as** follows: 0 to 14 min, 36.7% solvent  $B_1$ ; 14 to 34 min, 60% solvent  $B_1$ ; 34 to 41 min, 60% solvent  $B_2$ ; 41 min to end,  $90\%$  solvent  $B_2$ . Fractions were collected at 0.5- or 1-ml intervals and the radioactivity in each fraction was counted.

# **Lipid peroxidation**

Lipid peroxides were estimated as TBAR (24, 29). Preparations were diluted to a volume of 0.4 ml. Two ml of 0.25 N HC1 containing 0.375% **TBA** and 15% trichloroacetic acid were added together with 10  $\mu$ l of 5 mM  $\text{Fe}^{3+}$ as a catalyst to decompose lipid peroxides (29). The mixture was incubated at  $97^{\circ}$ C for 20 min, cooled, and centrifuged. The absorbance of the clear supernatant was measured at 532 nm. In some experiments, absorbance was converted to nmol (MDA) from a standard curve generated with **1,1,3,3-tetramethoxy-propane.** 

### **Prostanoids**

 $PGE_2$  and 6-keto- $PGF_{1\alpha}$ , the stable metabolite of PGI<sub>2</sub>, were identified by HPLC and estimated by a standard RIA procedure (23, 24). Antibodies were kindly supplied by Dr. Lawrence Levine (Brandeis University, Waltham, MA). The cross-reactivity of the PGE antibody was: 6-keto-PGF<sub>1 $\alpha$ </sub>, 0.4%; PGF<sub>1 $\alpha$ </sub>, 0.76%; PGF<sub>2</sub>, 0.31%;  $PGD<sub>2</sub>$ , 0.051%; AA, 0.00045%. The cross-reactivity of the 6-keto-PGF<sub>1 $\alpha$ </sub> antibody was: PGE<sub>2</sub>, 0.15%; PGD<sub>2</sub>, 0.02%; PGF<sub>2 $\alpha$ </sub>, 0.10%; AA, 0.005%.

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### **Preparation and lipid peroxide content of LDL**

Preliminary studies from our laboratory (22) showed that LDL were oxidized slowly at  $4^{\circ}$ C and that the rate of oxidation increased when preparations were incubated at 37°C in 96% air-4% CO<sub>2</sub>. Oxidation was accompanied by the decrease and ultimate disappearance of the yellow carotenoid color that is characteristic of freshly prepared LDL **(6).** IM did not have any effect on LDL oxidation. Relative lipid peroxide levels were estimated by a sensitive TBA assay that used a  $Fe<sup>3+</sup>$  catalyst to decompose lipid peroxides (29) and lipid peroxide levels were found to increase with the length of the incubation time.

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Peroxide levels (nmol MDA/200  $\mu$ g cholesterol) were  $0.7 \pm 0.1$  for LDL dialyzed against 0.15 M NaCl and  $1.9 \pm 0.1$  when the dialyzed LDL were incubated until the yellow carotenoid color disappeared. Higher peroxide levels were attained by continuing the incubation for different time intervals after the disappearance of carotenoid color. LDL with peroxide levels equal to or greater than 0.7 are designated LDL<sub>OXID</sub> in the present investigation.

With  $LDL<sub>BHT</sub>$ , incubation did not destroy the carotenoid color and peroxide levels only reached  $0.2 \pm 0.03/200$  $\mu$ g cholesterol after incubation times exceeding 24 hr. We found previously that BHT prevented both lipid peroxidation and some decomposition of lipid peroxides to TBAR  $(24, 29)$ . However, TBAR in LDL<sub>OXID</sub> were only reduced 30% when 100 *pM* BHT was added at the end of the incubation period (data not shown). Thus, very low TBAR values in  $LDL_{BHT}$  showed that BHT prevented lipid peroxidation when it was added early in the isolation of LDL.

When  $LDL<sub>OXID</sub>$  (2.2 nmol MDA/200  $\mu$ g cholesterol) solutions were adjusted to a density of 1.10 g/ml and recentrifuged, all of the cholesterol was recovered in the top (LDL) fraction. Electrophoretic mobilities did not differ for  $LDL_{BHT}$  and  $LDL_{OXID}$  with 0.7 nmol MDA/200  $\mu$ g cholesterol. The relative electrophoretic mobility of  $LDL<sub>OXID</sub>$ , measured against  $LDL<sub>BHT</sub>$ , increased to 1.1 when the peroxide level reached 1.9 nmol MDA/200  $\mu$ g cholesterol.

LDL were always isolated from individual sera and these LDL preparations underwent oxidation at different rates. Data for the oxidation of three different sera are reported in **Fig. 1.** The carotenoid color disappeared more rapidly from LDL-C than either LDL-A or LDL-B during the incubation process. Although LDL were oxidized at different rates, it was possible to obtain LDL preparations with the same lipid peroxide content by incubating until the disappearance of carotenoid color and then varying the subsequent incubation time.

# RESULTS Lipid classes in LDL<sub>BHT</sub> and LDL<sub>OXID</sub>

Lipids were extracted from  $LDL_{BHT}$  (0.05 nmol MDA/ 200  $\mu$ g cholesterol) and LDL<sub>OXID</sub> (1.5 nmol MDA/200  $\mu$ g cholesterol) and neutral lipid classes were separated by TLC. Both LDL<sub>BHT</sub> and LDL<sub>OXID</sub> contained CE  $(R_f)$ 0.90), TG  $(R_f 0.56)$  and C  $(R_f 0.19)$  (Fig. 2). LDL<sub>OXID</sub> appeared to contain less TG than  $LDL_{BHT}$ . FFA  $(R_f)$ 0.20) were not found in either preparation. Lipid extracts from  $LDL_{BHT}$  did not stain for lipid peroxides.

Lipid extracts from LDL<sub>OXID</sub> contained four new spots *(Rf* 0.42, **0.40,** 0.37, and 0.26) (Fig. 2) and these spots stained for lipid peroxides. Autoxidized cholesteryl arachidonate contained lipid peroxides  $(R_f \ 0.43, \ 0.37,$ and 0.32) and autoxidized trilinolenin contained a major lipid peroxide  $(R_f 0.26)$  (data not shown). These lipid peroxides with  $R_f$  values similar to the lipid peroxides identified in  $LDL<sub>OXID</sub>$  suggested that  $LDL<sub>OXID</sub>$  contained oxidized CE and TG.

Phospholipid classes were also separated by TLC. Both LDL<sub>BHT</sub> and LDL<sub>OXID</sub> contained PC  $(R_f 0.60)$ , Sph  $(R_f 0.60)$ 0.45), and LPC  $(R_f 0.32)$  but the relative size of the LPC spot from LDLoxID was much larger than the LPC spot from LDL<sub>BHT</sub> (Fig. 2). A FFA standard  $(R_f 0.75)$  was distinguishable from other neutral lipids including autoxidized cholesteryl arachidonate and autoxidized trilinolenin  $(R_f 0.96)$  by the phospholipid solvent system. A NL spot was present but a FFA spot was absent from the TLC plates obtained with lipids from LDLBHT and LDLOXID. Phospholipid classes did not stain for lipid peroxides, but



**Fig.** 1. LDL from individual sera are oxidized at different rates. Three LDL preparations (A, B and C) were isolated at the same time from three different sera, dialyzed against 0.15 **M** NaC1, sterilized, and then incubated at 37°C in 0.15 M NaCl. TBAR were measured at incubation times of 5, 10, and 15 hr and are reported as nmol MDA/200  $\mu$ g cholesterol.



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Fig. 2. Neutral lipids from LDL<sub>OXID</sub> (lane 1) and LDL<sub>BHT</sub> (lane 2) contain CE, TG, and C. LDL<sub>OXID</sub> contains four additional spots (a, b, c, d) that stained for lipid peroxides (data not shown). Phospholipids from LDL<sub>OXID</sub> (lane 3) and LDL<sub>BHT</sub> (lane 4) contained NL, PC, and Sph. LDL<sub>OXID</sub> also contained LPC and relatively less PC than LDL<sub>BHT</sub>. NL from LDL<sub>OXID</sub> stained for lipid peroxides (data not shown). Lipids were extracted by chloroform-methanol, separated by TLC, and visualized with iodine vapor. Spots were identified from the  $R_f$  values for authentic samples. See text for  $R_f$  values.

the neutral lipid spot ( $R_f$ 0.96) from LDL<sub>OXID</sub> stained for lipid peroxides. These data showed that oxidation released fatty acids from PC resulting in the formation of increased amounts of LPC. The fatty acids were evidently converted to derivatives that did not separate as FFA on TLC.

The absence of a FFA spot on TLC suggested that FFA derivatives were formed from the FFA released during LDL oxidation. Labeled **AA (0.4** *pM)* was added to a fresh LDL solution containing 7400  $\mu$ g cholesterol/ml and this mixture was oxidized, extracted with acidified ethyl acetate, and the labeled products were separated by HPLC. The extract contained free **AA** and small amounts of a number of labeled compounds which eluted as broad peaks **(Fig. 3).** Thus HPLC showed that free AA was converted to a number of derivatives that could explain the absence of FFA in  $LDL<sub>OXID</sub>$  when small amounts of FFA are released.

### **Immunoreactive products of LDL oxidation**

Since prostanoids are synthesized through cyclic endoperoxides and lipid autoxidation forms a number of cyclic and acyclic peroxides **(30, 31),** the cross-reactivities of LDLOXID containing FFA oxidation products were mea-

sured with antibodies to  $PGE_2$  and 6-keto-PGF<sub>la</sub>. LDL<sub>BHT</sub> did not cross-react with antibodies to either prostanoid (data not shown). Cross-reactivity to  $PGE<sub>2</sub>$  antibodies varied directly with the initial degree of lipid peroxidation in the  $LDL<sub>OXID</sub>$  preparations that were subsequently incubated for 24 hr in tissue culture media prior to measurement of PGE, levels by RIA **(Fig. 4).** Furthermore, immunoreactive materials were not the result of enzymatic prostanoid synthesis since a high concentration of IM had no effect on cross-reactivity (22). Lipid oxidation products showed unexpected specificity in that even highly oxidized LDL did not cross-react with antibodies to  $5ph$  **Sph**  $6-$  **keto-PGF<sub>1** $\alpha$ **</sub> (Fig. 4).** 

An add-back experiment was performed to show that with their binding.  $PGE<sub>2</sub>$  was measured in media containing  $LDL<sub>OXID</sub>$  (2.7 nmol MDA/200  $\mu$ g cholesterol) and media containing  $PGE<sub>2</sub>$  generated by incubating cell cultures with AA  $(22-24)$ . Media containing  $LDL<sub>OXID</sub>$  (91) pg apparent PGE,) were combined with AA media (88 pg de novo PGE,) and PGE, levels were determined for these **LPC LEC LECC LECC LECC LECC LECC LECC LECC LECC** 



Fig. 3. [U<sup>-14</sup>C]AA is converted to a number of derivatives during cooxidation with LDL. Labeled **AA (0.4** *p~)* and LDL **(7400** pg cholesterol/ml) were co-oxidized to 3.7 nmol MDA/200 µg cholesterol. Labeled derivatives were extracted in acidified ethyl acetate and separated by HPLC using mixtures of **acetonitrile-isopropanol-aqueous**  phosphoric acid.



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Fig. 4. Cross-reactivity of LDL<sub>OXID</sub> to PGE<sub>2</sub> antibodies varies directiy with lipid peroxide content and is not inhibited by IM. LDL<sub>OXID</sub> does not cross-react significantly with 6-keto-PGF<sub>1</sub> antibodies. LDL from 17 individual sera were prepared with different levels of lipid peroxidation. LDL concentrations **wre** adjusted to *800* pg cholesterol/ml and the LDL were incubated in tissue culture media alone for **24** hr at **37OC** in the absence or presence of **fO** *pM* **1M.** RIA **was**  then used to estimate prostanoid levels. PGE<sub>2</sub> in the graph refers to immunoreactive PGE<sub>2</sub>-like material.

mixtures. The PGE<sub>2</sub> level measured in the mixture was 105% of the PGE<sub>2</sub> level calculated from values obtained for the separate samples. These data showed that  $LDL<sub>OXID</sub>$ did not interfere with the determination of PGE<sub>2</sub>.

Cross-reactivity to  $PGE<sub>2</sub>$  antibodies was a function both of the initial degree of lipid peroxidation and of the incubation time. For example, 24-hr incubations of seven different LDL<sub>OXID</sub> preparations adjusted to 800  $\mu$ g/ml in tissue culture medium produced significant increases in immunoreactive PGE<sub>2</sub> (0.040  $\pm$  0.002 nmol/plate at 0 hr compared to  $0.088 \pm 0.003$  nmol/plate at 24 hr). We also extended the incubation time period to 72 hr. Data for two representative LDL<sub>OXID</sub> preparations incubated for 72 hr in tissue culture media alone are compared in Fig. 5. Cross-reactivity to the PGE<sub>2</sub> antibody varied directly with the initial lipid peroxide levels in the  $LDL<sub>OXID</sub>$ , and large increments in immunoreactive materials occurred during incubation.

Lipid peroxide levels were also measured during the experiments described in Fig. 5. In contrast to crossreacting material, MDA levels (absorbance at 532 nm) did not increase with time when  $LDL<sub>OXID</sub>$  were incubated in tissue culture media. These data showed that tissue culture media contained sufficient amounts of antioxidants to prevent further lipid peroxidation. Despite the absence of further lipid peroxidation, the increase in cross-reacting material over time occurred even in the presence of 100  $\mu$ M BHT. In this context, after 24 hr incubation there was no significant difference in immunoreactive  $PGE<sub>2</sub>$  in incubations with or without BHT  $(0.084 \pm 0.005 \text{ nmol/plate compared to } 0.088 \pm 0.003$ nmol/plate). These experiments showed that preformed lipid peroxides in the LDL<sub>OXID</sub> were converted during incubation in tissue culture media alone to products that expressed immunoreactivity toward PGE<sub>2</sub> antibodies.

### Solubility of immunoreactive products from LDL<sub>OXID</sub>

A LDL<sub>OXID</sub> solution was adjusted to d 1.10  $g/ml$  with KBr and LDL was re-isolated by ultracentrifugal flotation. The contents of the centrifuge tube were separated into top, middle, and bottom fractions and each fraction



Fig. 5. Cross-reactivity of LDL<sub>OXID</sub> to PGE<sub>2</sub> antibodies varies directly with both the initial lipid peroxide content and the incubation time. Lipid peroxide levels decrease or remain constant during the incubation period. Two LDL<sub>OXID</sub> preparations were adjusted to 800 µg cholesterol/ml and incubated in tissue culture media alone for 0, **24, 48,**  and **72** hr at **37OC.** Lipid peroxides were measured as TBAR *(0,* **A)** and immunoreactive PGE<sub>2</sub>-like material was measured by RIA ( $\circ$ ,  $\wedge$ ) after each time interval. The initial lipid peroxide content of one LDL<sub>OXID</sub> preparation *(0)* was greater than the initial lipid peroxide content **of** the other  $LDL<sub>OXID</sub>$  preparation ( $\triangle$ ). Each data point in the graph was generated from two or three separate incubations of the LDL<sub>OXID</sub> and is reported as mean  $+$  SEM.

was analyzed for cholesterol and immunoreactive products. All of the cholesterol was recovered in the top fraction. Apparent PGE<sub>2</sub> levels (ng/ml) were 33 before centrifugation and **32, 33,** and 47 in the top, middle, and bottom fractions, respectively. These data showed that immunoreactive products were not bound to  $LDL<sub>OXID</sub>$ since they were not concentrated in the  $LDL<sub>OXID</sub>$  fraction by ultracentrifugation.

# Immunoreactive **products** *of* **I1** oxidation

The cross-reactivity to  $PGE_2$  antibodies of Il, a synthetic intravenous fat emulsion, was measured to ascertain whether cross-reactivity was a general result of lipid oxidation or the specific result of LDL oxidation. I1 did not contain significant amounts of lipid peroxides and I1 did not generate lipid peroxides during incubation in 96% air-4% **COz** at **37°C.** However, large amounts of lipid peroxides were formed when 10  $\mu$ M  $\text{Fe}^{3+}$  was added to the incubation solution (Table **1).** I1 showed a low level of cross-reactivity to  $PGE_2$  antibodies and cross-reactivity was not increased in  $II_{\text{OXID}}$  (Table 1) even though MDA levels in Il<sub>OXID</sub> were comparable to MDA levels in  $LDL<sub>OXID</sub>$  (Fig. 4). The Fe<sup>3+</sup> catalyst did not affect crossreactivity since both MDA levels and cross-reactivities were enhanced when LDL was oxidized in media containing 10  $\mu$ M  $\text{Fe}^{3+}$  (data not shown). The differences between

**TABLE** 1. **Free AA and oxidation enhance the cross-reactivity** of Il to PGE<sub>2</sub> antibodies

Incubation Time	<b>MDA</b>	Apparent PGE <sub>2</sub>
hт	nmol/plate	pmol/plate
А. Il		
12	0.2	$8.6 \pm 0.15$
24	0.2	$12.3 \pm 0.37$
48	0.2	$10.1 \pm 1.00$
$I1 + AA$		
12	0.8	$9.5 \pm 0.37$
24	0.2	$12.2 \pm 0.22$
48	0.4	$10.3 \pm 0.81$
B. Il <sub>oxid</sub>		
12	7.7	$14.9 \pm 0.18$
24	12.5	$14.5 + 0$
48	10.1	$13.6 \pm 0$
$I_{QXID} + AA$		
12	7.1	$16.0 \pm 0.22$
24	12.5	$24.2 \pm 1.41$
48	10.1	$28.0 \pm 0.15$

Il samples with and without  $120 \mu M$  AA were incubated in saline alone (11) or saline containing  $10 \mu M$   $Fe^{5+}$  ( $II_{\text{OXID}}$ ) for different time intervals and MDA was measured. Il or IL<sub>OXID</sub> content was adjusted to 760  $\mu$ g/ml **with tissue culture media and incubation was continued for 24 hr.** Two **separate incubations were performed for each time point. Apparent PGE, was then measured (RIA) and reported as mean f SEM. Data were subjected to a two-way analysis of variance (see text).** 

Since I1 does not contain either free or bound AA, I1 was mixed with free AA and cross-reactivity to  $PGE<sub>2</sub>$ antibodies was measured before and after lipid peroxidation (Table 1). Oxidation and AA effects were compared by a two-way analysis of variance and this analysis showed a strong interaction between free AA and oxidation in enhancing cross-reactivity (F 9.92, *P* 0.005). Free AA also increased cross-reactivity when it was added during LDL oxidation (data not shown). These results showed that the co-oxidation of free AA during both I1 and LDL oxidation generated cross-reacting material.

# DISCUSSION

Newly isolated LDL contain very little lipid peroxide but peroxides are readily formed during dialysis against 0.15 **M** NaCl and these preparations, which contain  $0.7 \pm 0.1$  nmol MDA/200  $\mu$ g cholesterol, must be considered as  $LDL<sub>OXID</sub>$ . Oxidation is blocked by BHT and it is probable that  $LDL_{BHT}$  more nearly represents native LDL than  $LDL<sub>OXID</sub>$ . Although the BHT experiments show that much of the native LDL that are present in plasma are not oxidized, these experiments do not eliminate the possibility that LDLOXID are formed in plasma and other tissues and then taken up more rapidly than unoxidized LDL by cells (11-16).

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Dialyzed  $LDL<sub>OXID</sub>$  are capable of more extensive lipid peroxidation in which the TBAR content is increased to as much as 6 nmol MDA/200  $\mu$ g cholesterol. Oxidation is controlled by incubating  $LDL<sub>OXID</sub>$  under 96% air-4% **COz** at **37OC** for various time intervals and oxidation under these conditions is much slower than oxidation catalyzed by Cu<sup>2+</sup>. The rate of mild oxidation is actually a sensitive measure of the susceptibility to autoxidation of LDL preparations from different human plasma samples. Rate studies on autoxidation show that it may be possible to generate circulating LDL with different peroxide levels and indeed plasma and isolated plasma lipoproteins from normal and diseased subjects contain different concentrations of lipid peroxides **(4, 7-10).** 

 $LDL<sub>OXID</sub>$  contain several lipid peroxides that are confined to esterified neutral lipids. Thus lipid extracts from  $LDL<sub>OXID</sub>$  contained lipid peroxides that migrated as discrete spots with  $R_f$  values in the same region as spots from autoxidized cholesteryl arachidonate and trilinolenin. Lipid peroxides were not identified either in the region of the TLC plate occupied by simple fatty acid hydroperoxides or the phospholipid band that remained at the origin.

Other investigators found that the PC fraction decreased and the LPC fraction increased when LDL were oxidized extensively by cells or **Cu2+ (13, 16, 20,** 21) and dation in the present study. The formation of increased LPC was shown in those studies to be the result of enhanced phospholipase  $A_2$  activity. Since FFA and simple fatty acid hydroperoxides were not detected on TLC, FFA were evidently converted to a number of complex oxidation products such as the oxidation products that were separated by HPLC after the co-oxidation of LDL and labeled AA.

Autoxidation forms a number of acyclic and cyclic lipid peroxides **(30, 31)** which have different properties in LDL<sub>OXID</sub>. Lipid peroxides in esterified neutral lipids such as cholesteryl esters and triglycerides are evidently the source of TBAR, since fatty acid hydroperoxides were not identified. The susceptibility of cholesteryl esters in LDL to lipid peroxidation is interesting in that early investigators identified fatty acid oxidation products in cholesteryl esters from aortas of subjects with severe atherosclerosis **(32).** The accumulation of these lipids in plaques may provide indirect evidence for circulating LDLOXID.

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Complex oxidation products that cross-react with antibodies to  $PGE_2$  are not unique to  $LDL<sub>OXID</sub>$ . Similar material is generated when the synthetic fat emulsion I1 is co-oxidized with free AA. However, significant amounts of the immunoreactive material are not generated during the oxidation of I1 alone. FFA released from PC during LDL oxidation are evidently converted to unbound oxidation products that cross-react specifically with antibodies to PGE<sub>2</sub>.

**A** number of studies have indicated that LDL stimulate  $PGE<sub>2</sub>$  synthesis and either stimulate or inhibit  $PGI<sub>2</sub>$  synthesis (8, **9, 33-37).** Prostanoids are usually measured by RIA and the formation of material cross-reacting with PGE<sub>2</sub> but not 6-keto-PGF<sub>1 $\alpha$ </sub> (the stable derivative of PGI<sub>2</sub>) that we report here may affect the interpretation of data from other published studies.

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